Methanol and the Optically Pumped Far-Infrared Laser

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Abstract—New results on the generation and spectroscopic analysis of optically pumped far-infrared (FIR) laser emission from CH $_3$ OH have been obtained as part of a systematic study of methanol isotopomers as FIR laser sources utilizing the extended line coverage available from a recently developed high-resolution CO $_2$ laser of high efficiency. For normal CH $_3$ OH, six new shortwavelength lines have been found using a 2 m long Fabry–Perot FIR laser cavity. Accurate heterodyne frequency measurements are reported for 14 CH $_3$ OH FIR laser lines, nearly all above 100 cm $^{-1}$, as well as accurate frequency offsets for most of the corresponding CO $_2$ pump lines. Spectroscopic assignments are presented for nine high-frequency FIR laser lines in four pump systems.

I. INTRODUCTION

E report new results on the generation and spectroscopic analysis of optically pumped far-infrared (FIR) laser emission from CH₃OH. This paper is a major part of our systematic investigation of methanol isotopomers as FIR laser media utilizing a newly designed FIR laser system. The system combines several different FIR laser cavity designs with the extended line coverage available from a recently developed high-resolution CO2 laser of high efficiency. With this system, new FIR laser lines pumped by CO2 fundamental band lines of high J, CO₂ sequence band lines, and CO₂ hot-band transitions have been observed. The new FIR laser lines extend over a broad spectral region and contribute significantly to the frequency coverage of the optically pumped FIR laser. In the program so far, over 150 new lines have been observed for eight isotopic species of methanol (CH₃OH, ¹³CH₃OH, CD₃OH, ¹³CD₃OH, CD₃OD, ¹³CD₃OD, CH₃OD, and CH2DOH), and 76 of the new lines plus 23 previous lines with only wavelength observations have been measured in frequency by accurate heterodyne techniques.

Almost immediately after the first successful generation of FIR laser radiation by optical infrared (IR) pumping with a CO₂ laser 25 years ago, methanol and its isotopomers were

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recognized as important sources because of their richness of IR pump absorptions, output power, and breadth of FIR spectral coverage. Up to now, more than 570 lines have been observed for the parent ¹²CH₃¹⁶OH species alone and about 2000 lines in total for all of the isotopomers of methanol including ¹³CH₃OH, CH₃¹⁸OH, CD₃OH, ¹³CD₃OH, CD₃OD, ¹³CD₃OD, CH₃OD, CH₂DOH, CH₂DOD, and CHD₂OH [1]-[4]. The observed methanol FIR laser lines cover a wavelength range from 19-3030 μ m, corresponding to 526-3.3 cm⁻¹ in wavenumber units or 16 THz to 99 GHz on the frequency scale. Therefore, methanol FIR lasers play an important role in bridging the full gap of radiation between the microwave and optical regions and have been widely used in a variety of research fields, such as laser magnetic resonance (LMR) spectroscopy, tunable FIR spectroscopy (TuFIR), and plasma diagnostics.

Several features contribute to the success of methanol isotopomers as excellent FIR laser media [5]. The large permanent dipole moments along both a and b axes lead to a strong FIR spectrum and liberal transition selection rules. The complexity of the energy level structure due to the large-amplitude torsional motion gives rich FIR and IR spectra, and the methanol IR absorption bands overlap well with the CO_2 laser bands. The favorable separation of the torsional energy states gives the possibility for high-frequency (short wavelength) torsional FIR lasing transitions; apart from the methanol isotopomers, only CH_2F_2 and N_2H_4 are presently known to be comparably prolific sources of laser lines above 2 THz. It is these characteristics of the methanol isotopomers that contribute greatly to the size of the FIR laser line catalog.

This paper deals with our new CH₃OH FIR laser data and is organized as follows: Section II discusses the energy structure and the low resolution IR spectra of the methanol isotopomers; Section III presents the main features of the CO₂ pump laser and the characteristics of the two FIR laser cavities used in this paper; and Section IV describes and comments on the new results that we have obtained on FIR laser line observations and spectroscopic assignments for CH₃OH.

II. METHANOL ENERGY STRUCTURE AND IR SPECTRA

Methanol is a slightly asymmetric molecule possessing a large-amplitude internal rotation or torsional motion with a threefold torsional potential and a plane of symmetry. The tunneling motion of the OH framework with respect to the ${\rm CH_3}$ top lifts the torsional degeneracy and splits the levels into A and E symmetry species. The torsional energy can be

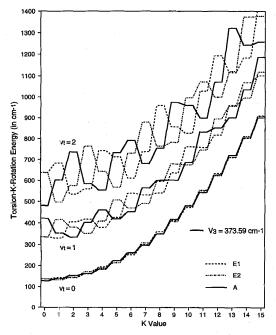


Fig. 1. Torsion-K-rotation energy level structure for the lowest three torsional states of CH₃OH. The height V_3 of the top of the barrier to internal rotation is also indicated.

approximately expressed as

$$E_{tor} = FP_{\gamma}^2 + \frac{1}{2}V_3(1 - \cos 3\gamma),$$
 (1)

where F is the reduced rotational constant equal to the inverse of the reduced moment of inertia of the top (CH₃) and framework (OH), P_{γ} is the torsional angular momentum conjugate to the torsional angle γ , and V_3 is the torsional potential barrier (~373 cm⁻¹ for the normal species). The first excited torsional levels ($v_t=1$) straddle the top of the barrier. This torsional energy is superimposed on top of the zero-order rotational energy

$$E_{rot} = \frac{1}{2}(B+C)J(J+1) + [A - \frac{1}{2}(B+C)]K^2, \quad (2)$$

where A, B, and C are the rotational constants, and K is the projection of the overall angular momentum J along the a-axis parallel to the methyl top axis. The J-independent part of the torsion-K-rotation energy for the $v_t=0$ -2 levels is displayed in Fig. 1. The complex structure is very broadly spread in energy, explaining the success of methanol as a lasing medium in covering such a wide range of wavelengths through transitions both within and across the different torsional states.

Both IR and FIR spectra of methanol are very rich due to this torsional complexity and the liberality of the selection rules summarized in Table I. Transitions can occur within a K energy ladder with $\Delta K = \Delta v_t = 0$ (a-type) or between different K stacks with $\Delta K \geq 1$ and no restriction on the change Δv_t of torsional quantum number (b-type). The same selection rules apply to IR transitions between different vibrational states.

Low-resolution and normal mode studies [6]–[7] show that several vibrational bands are located in the region of the 9 and 10 μ m CO₂ laser bands. The overlap between the IR

TABLE I
SUMMARY OF TRANSITION SELECTION RULES
AND TRANSITION STRENGTHS FOR METHANOL

Туре	ΔΚ	Δv _t	ΔJ	Branch		Transition Strength Sa							
a-type			0	Q	$A^{\pm} \leftrightarrow A^{-/+}$	(2J+1)K ² /J(J+1)							
(μ_a)	0.	0	+1	R	$A^{\pm} \leftrightarrow A^{\pm}$	[(J+1) ² -K ²]/(J+1)							
			-1	P	$A^\pm \leftrightarrow A^\pm$	(J ² -K ²)/J							
			0	Q	A [±] ↔ A-/+	(1/4) <k±1 k>2(2J+1)[1-K(K±1)/J(J+1)]</k±1 k>							
	±1	even	even +1	R	$A^{\pm} \leftrightarrow A^{\pm}$	$(1/4) < K \pm 1!K > 2(J+1 \pm K)(J+2 \pm K)/(J+1)$							
b-type			-1	P	$A^{\pm} \leftrightarrow A^{\pm}$	$(1/4) < K \pm 1 K > 2(J - / + K)(J - / + K - 1)/J$							
(μь)										0	Q	$A^{\pm} \leftrightarrow A^{\pm}$	$(1/4) < K \pm 1 K > 2(2J+1)/[1-K(K\pm 1)/J(J+1)]$
	±1	odd	+1	R	$A^{\pm} \leftrightarrow A^{-/+}$	$(1/4) < K \pm 1 K > 2(J+1 \pm K)(J+2 \pm K)/(J+1)$							
			-1	_ P	$A^{\pm} \leftrightarrow A^{-/+}$	$(1/4) < K \pm 1 K > 2(J - / + K)(J - / + K - 1)/J$							

a <K'lK"> denotes a torsional overlap matrix element

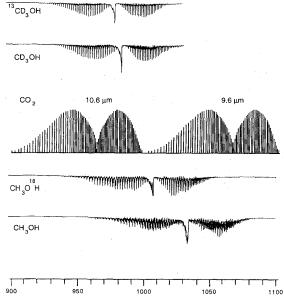


Fig. 2. Superimposed CO-stretching bands of several isotopomers of methanol and the ${\rm CO_2}$ laser spectrum.

absorption bands and the CO_2 laser spectrum is excellent, and more than 100 pump coincidences between CH_3OH absorptions and CO_2 laser lines have now been observed. Exchange of H by D, of an ordinary ^{12}C carbon atom by the ^{13}C isotope, or of ^{16}O by ^{18}O in CH_3OH shifts the vibrational bands somewhat, but the IR absorption still falls within the CO_2 laser spectrum as shown in Fig. 2. However, the shifts in molecular rotational and vibrational levels with isotopic substitution are sufficient to generate entirely new sets of pump coincidences and FIR laser lines. Table II lists the vibrational assignments for the various isotopomers of methanol in the CO_2 laser spectral region, collected from [6] and [7]. Note that combination and hot bands are not considered in this table.

III. OPTICALLY PUMPED FIR LASERS

In the general scheme of an optically pumped methanol FIR laser, molecules are pumped to a certain excited vibration-rotation-torsion level $(v_t\tau K,J)^v$ by a selective CO₂ pump laser line which is in coincidence with a molecular IR absorption line. (In our notation, v_t is the torsional quantum number, τ labels the torsional symmetry, K is the projection of total angular momentum J along the a-axis, and v is the vibrational

TABLE II
LOW-RESOLUTION VIBRATIONAL ASSIGNMENTS
FOR METHANOL ISOTOPOMERS ([6] AND [7])

Molecule		A'			A"			
*********	v(CO)	ρ(CH ₃)	δ(CH ₃)-s	δ(CH ₃)-a	ρ(CH ₃)	δ(CH ₃)-a		
CH ₃ OH	1033.5	1074.5			1145.0			
¹³ CH ₃ OH	1018.1	1068.1			[1144.0]			
CD ₃ OH	984.4		1130.6	1068.2	897.5	1068.2		
13CD3OH	980.2		1111.5	[1059.0]	[878.2]	[1060.3]		
CH ₃ OD	1038.4				1142.0			
CD ₃ OD	981.8	1070.5	1134.3	1027.9	895.0	1070.5		
13CD3OD	974.9	1064.5	1113.6	1024.0	[878.2]	[1060.3]		

(The methanol symmetry is C_s , hence the 12 normal vibrations are represented by T = 8A' + 4A''.

s - symmetric, a - antisymmetric. Wavenumbers in brackets are calculated from force constants.)

quantum number.) Population inversion is then achieved relative to lower rotational levels in the excited vibrational state, where the population is very small before the optical pumping, or sometimes in the ground vibrational state due to depletion of the lower level population by the optical pumping. Stimulated emission can thus occur between those levels with inverted populations. The FIR laser lines can be of either a or b type, as given by the selection and intensity rules in Table I, and a typical pumping scheme and the possible FIR emissions are shown in Fig. 3. The dashed arrows indicate allowed but weak FIR transitions that do not always lase. So far, most observed transitions belong to three-level systems, shown schematically in Fig. 4(a) and (b), where the lasing occurs directly from the upper level into which the pump energy is deposited. For the three-level system depicted in Fig. 4(a), the pumping is from the ground state to the excited state and the FIR lasing takes place within that same state or to a nearby vibrational state. More recently, a new scheme in which the pump coincides with a $v = 2 \leftarrow 1$ hot band transition was proposed and is shown in Fig. 4(b) [8]. Many of the systems are associated with the important Fermi or Coriolis perturbation mechanisms [9] so provide spectroscopic information that is invaluable for understanding interactions among the excited vibrational states. Currently, however, over half of the FIR lasing transitions still remain unidentified in spite of much spectroscopic effort. We thus speculate that FIR laser action might occur by the four-level process shown in Fig. 4(c). where energy and population both flow out of the upper state prepared by the IR pumping and along an interaction channel over to another state from which the FIR lasing takes place at a later time. We are actively seeking spectroscopic confirmation of this type of process, which would represent a significant new technique for the study of intramolecular vibrational energy redistribution (IVR) mechanisms in methanol.

The FIR radiation is linearly polarized either parallel or orthogonal to the pump radiation depending on the sum of ΔJ_{pump} and $\Delta J_{\rm FIR}$, the J-value changes in the pump and FIR laser transitions. The rule for the relative polarization of the pump and emission fields was stated by Henningsen [10] as

$$\Delta J_{pump} + \Delta J_{\text{FIR}} = \begin{cases} \text{even} & \Longrightarrow & \parallel \\ \text{odd} & \Longrightarrow & \perp \end{cases}$$
 (3)

Polarization information from the observed FIR laser emissions often gives clues to their identification. As indicated in

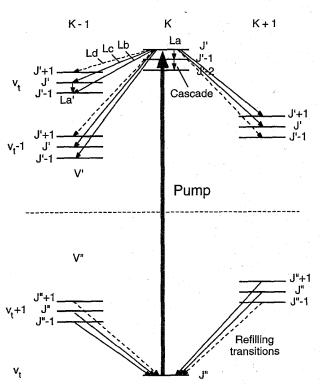


Fig. 3. Schematic diagram of IR-pump/FIR-laser energy level scheme with possible FIR laser transitions.

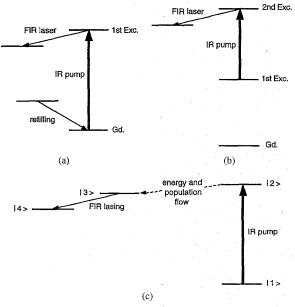


Fig. 4. Possible IR-pump/FIR-laser transition scheme with (a) Pumping in the vibrational fundamental band. (b) Pumping in the hot band. (c) Indirect pumping via a speculative four-level mechanism in which energy and population both flow from pumped level $|2\rangle$ to lasing level $|3\rangle$ along interaction channels coupling the levels.

Fig. 3, a triad of emission lines whose wavenumbers obey the approximate combination relation

$$L_a + L_c - L_b \approx 0 (4)$$

may be observed. In the triad, L_a and L_b have the same

 $\Delta J_{\rm FIR}$ and therefore have the same polarization, while L_c , the Q-branch transition in the excited state, has the opposite polarization from L_a and L_b . Observation of a triad greatly aids the FIR laser assignment since the a-type laser emission frequency gives J information for the upper pumped level from the approximate relation

$$J_{upper} \approx \frac{L_a}{2B'},$$
 (5)

where B' is the effective excited state rotational constant. The polarization of L_c then determines ΔJ_{pump} , and the L_c wavenumber can often give the other quantum numbers $(v_t \tau K)$ since the excited state energy structure is generally not too different from the known ground state levels in the absence of perturbations.

As seen in the literature, searches for FIR laser emission from various molecular gases are often aimed at uncovering the excited-state energy structure of the molecule in question. In our case, the observation and identification of FIR laser emissions have provided important confirmation of the IR assignments in a number of doubtful cases where line series were perturbed or strongly blended. The traditional IR assignment techniques often lose their power in the very condensed C-O stretch band, particularly for medium or high J where line overlap in the congested J-multiplets of the R and P subbranches can be very serious.

The experimental work was carried out at the Time and Frequency Division, National Institute of Standards and Technology, in Boulder. The optically pumped FIR laser system used in this work consisted of a CO₂ pump laser and two different FIR laser cavity designs that will be briefly discussed below. The CO₂ pump laser had a 1.5 m cavity and a ribbed discharge tube that increases the effective resolution of the grating in the cavity by preventing wall bounces and thereby eliminating waveguide modes [11]. Output coupling in the laser is derived in zero order from specially blazed gratings of 135, 150, 163, or 171 lines/mm. The zero-order output coupling from the 135 line/mm grating was measured to be approximately 5%, and the others behaved comparably. Altogether, this CO₂ system lases on more than 250 lines including the regular 9.4 and 10.4 μm bands from J=0to J=62, many of the 10.8 μm hot-band lines, many of the 10 μ m sequence-band lines, some of the 9 μ m sequence lines, and a substantial number of the new 9 μ m hot-band lines, with typical powers of order 25 W for the regular lines, 11 W for the hot-band lines, and 6 W for the sequence band lines [12].

Two FIR laser cavities were used in this paper. One was a 2 m long metal-dielectric rectangular waveguide cavity pumped longitudinally through a 1 mm hole in a flat copper end mirror and is described in detail in [13]. The other was a low-loss (<0.5% diffraction loss for wavelengths below 150 μ m) FIR Fabry-Perot cavity consisting of a 2 m long Pyrex tube of 36 mm diameter closed by a fixed copper flat mirror and a gold-coated concave mirror with 4 m radius-of-curvature and operated in a folded confocal geometry [14]. In both cavities, the end mirror was attached to a precision micrometer in order to tune the cavity into resonance with the FIR laser modes. The FIR laser radiation

was coupled out of the cavity by a 45° copper mirror and detected by a metal-insulator-metal (MIM) diode or a pyroelectric detector. The amplitude of the detector signal was recorded as an estimate of the relative intensity of each FIR laser line. A quartz disc was frequently placed in front of the MIM diode in order to block unwanted CO₂ laser carrier.

In searching for new FIR laser lines, the first step was to monitor the optoacoustic response of the laser gas to the chopped CO₂ pumping radiation by means of a microphone placed inside the FIR laser cavity, in order to find CO2 lines having good coincidences with molecular absorptions. The FIR laser cavity length was then tuned into resonance with the laser modes. The FIR laser wavelength was determined to within an uncertainty of approximately $\pm 0.5 \mu m$ by measuring the distance travelled by the micrometer for 10 or more wavelengths. For most of the new FIR laser lines, the frequencies were measured to higher precision by using the heterodyne technique described in [15]. In this technique, two frequencystabilized CO2 lasers are locked to the Lamb-dips at their respective line centers. The beams from these lasers, which have frequencies $\nu_{co2}(I)$ and $\nu_{co2}(II)$ and moderate power, are mixed on a MIM diode with the unknown FIR laser radiation u_{FIR} plus a microwave signal $u_{\mu\,wave}$ supplied in the 2-18 GHz range from a synthesized signal generator. A spectrum analyzer is then employed to search for a resulting beat note at the frequency given by the equation:

$$\nu_{\text{FIR}} = |n_1 \nu_{co2}(I) - n_2 \nu_{co2}(II)| \pm m \nu_{\mu \, wave} \pm \nu_{beat},$$
 (6)

where n_1 , n_2 , and m are integers corresponding to the respective harmonics generated in the MIM diode, and ν_{beat} is the beat frequency. The CO₂ frequencies and harmonic orders are chosen to give a beat note below 1.5 GHz. The intensity of the beat signal decreases as the harmonic orders increase; hence, we generally aim to satisfy the optimum condition $n_1 = n_2 = m = 1$. However, when measuring the higher frequency ($\lambda < 50 \mu m$) methanol lines, this was not possible. For the 26.2 μ m line at 11.4 THz [16], for example, we obtained the beat note with $n_1 = n_2 = 3$ and m = 1. The beat note is amplified and displayed on a spectrum analyzer, using a peak-hold feature that records the peak signal as the FIR laser is tuned over its gain curve. The center frequency of the gain curve is then measured with a marker derived from an accurate oscillator. The estimated 1σ accuracy of the FIR laser frequency measurements is $\Delta \nu / \nu = 2 \times 10^{-7}$, due mainly to the precision of the resettability of the FIR laser.

With the rich spectral output available from the efficient, high-resolution CO_2 laser [11], [17] we found it important to measure the precise IR pump offset for each FIR laser line in order to be absolutely sure of the identity of the CO_2 pump transition. When the latter was located in a dense region of the CO_2 laser spectrum, one could have regular, hot-band and sequence-band laser lines all crowded closely together with significant output powers and not readily distinguishable simply from the setting of the CO_2 laser tuning micrometer. The offset measurements were performed by adjusting the system to the peak of the FIR emission and then beating the CO_2 pump laser frequency ν_{pump} against a second Lamb-dipstabilized CO_2 reference laser frequency ν_{ref} in a MIM diode.

				-	•				
CO ₂ Pump	VCO2	Offset ^a	FIR Laser Line			Rel.	Pressure	Rel Int	Ref.
	[cm ⁻¹]	[MHz]	λ [μm]b	ν [cm ⁻¹]c	v _{FIR} [MHz] ^d	Pol.	[Pa]	[mV]	
10R (04)	964.76898		57.4				16.5		new
10R (32)	983.25225	-48	48.283	207.11286	6 209 087.2	1	32.0	0.5	new
10R (46)	990.61963	+48	41.630	240.21218	7 201 380.0	1	40.0	0.1	14
		+48	50.389	198.45711	5 949 594.6	1	32.0	0.3	14
		(+15)	52.004	192.29392	5 764 826.7	丄	32.0	0.1	2, 4
		+45	61.819	161.76189	4 849 499.5	1	22.7	0.2	14
		(+15)	64.156	155.87022	4 672 871.5		53.3	0.1	2, 4
		+46	80.747	123.84435	3 712 760.2	ı	32.0	0.3	14
10R (48)	991.56580	-16	69.188	144.53396	4 333 019.1	H	21.3	0.2	14
		-18	97.981	102.06043	3 059 694.8	1	32.0	0.5	2, 4
10R (50)	992.48480		102.061	97.98084	2 937 391.8	I	16.0	0.1	new
			130.6				17.3		new
10R (52)	993.37643	+40	49.694	201.23294	6 032 811.7	ı	20.0	0.2	14
		+34	60.676	164.81093	4 940 907.5	Τ	45.3	0.1	14
			104.3						new

168.08929

89.52265

5 039 190.0

2 683 821.6

TABLE III New Optically Pumped FIR Laser Lines of $\mathrm{CH}_3\mathrm{OH}$ with Frequency and Offset Measurements

59.492

111.704

-25

+35

9P (56) 1010.04283

1033.48800

9P (34)

For pumping with the regular CO_2 bands, the reference laser was locked to the same line as the main pump laser, and the beat frequency observed on the spectrum analyzer then gave the pump offset $|\nu_{pump} - \nu_{ref}|$. For sequence or hot-band pumping, the reference laser was locked to the closest regularband CO_2 line and an additional microwave frequency $\nu_{\mu\,wave}$ was mixed in to bring the beat note into the range of the spectrum analyzer. The offset was then obtained to an estimated experimental uncertainty of ± 2 MHz from the relation

$$\nu_{pump} = (\nu_{ref} \pm \nu_{\mu \, wave}) \pm \nu_{beat}. \tag{7}$$

IV. RESULTS AND DISCUSSION

Table III reports observations of six new FIR laser lines for CH_3OH along with 14 new frequency measurements of lines previously reported in the literature. The newly measured lines cover the frequency range from 2.6-7.2 THz.

With the IR and FIR data sets available for CH₃OH [18], [19], we have been able to assign the quantum numbers for the transitions associated with two of the new FIR laser pump systems and find new assignments for two additional previous systems. Nine of the FIR laser transitions have been identified, all high-frequency lines involving torsionally excited states. The results are given in Table IV and show excellent agreement between the accurate heterodyne FIR laser wavenumbers and those derived from combination difference relations based on our high-resolution spectroscopic data.

This type of result [20] has served as an extremely valuable check on the spectroscopic assignments, particularly for the weaker IR transitions in the torsionally excited subbands of the CO-stretching spectrum whose positions are significantly downshifted relative to the $v_t=0$ subbands by up to several units in J.

49.3

22.7

0.5

1.2

14

new

In the present paper, for example, we correct the assignment of the 9R(2) system pump transition reported in the earlier reviews as the A^+ component of the $R(113, 24)^{co}$ IR K-doublet [1], [20]. In actuality, the pump absorption is the A- component, which fact emerges clearly from our combination loops when we account for the substantial Kdoubling contribution to the energy of the lower (022+, 24)co level of the 285.24496 cm⁻¹ FIR laser line. We have also confirmed the reported P(123, 24)co pump for the 10R(46) + 48 MHz system [21] for which a misassignment existed in earlier literature [20], [22]. In the previous spectroscopic work leading to the correct identification, several high-frequency FIR laser lines were predicted from combination loop relations [21], and we have now observed and measured three of these, thus confirming the assignment. This confirmation is important because the single a-type FIR laser line observed earlier is not very sensitive to the $(v_t au K)$ quantum numbers so could not be identified unambiguously. Now, however, the observation and assignment of four high-frequency b-type laser lines leave absolutely no doubt about the identity of the P(123, 24)co

a Offsets in parentheses have not been measured but are literature values.

b Lines with wavelengths reported to one decimal place only have not been measured in frequency.

^c Calculated from frequency measurement with 1 cm⁻¹ = 29 979.2458 MHz.

d 1σ relative uncertainty is $\Delta v/v = 2 \times 10^{-7}$.

+	V	IR Absorptiona	FIR Laser Transitionb	Vohe	Re
Α	SSIGNMENTS	of New CH ₃ OH FII	R LASER LINES OPTICALLY PUMPS	ED BY A CO ₂	LASER
			TABLE IV		

CO ₂ Pump +	F F			ransition ^b	V _{obs}	Rel	V _{calc} ^b	
Offset [MHz]	[cm ⁻¹]	P/Q/R	$(v_t'\tau'K',J')^{\nu}$	\rightarrow	$(v_t"\tau"K",J")^{v"}$	[cm ⁻¹]	Polb	[cm ⁻¹]
9R(02) + 25	1066.03820	R(114,23)co	(114,24)co	\rightarrow	(023,24)co	236.59015	Τ	236.5901
				\rightarrow	(023,23)co	274.78829	ı	274.7879
9R(02) - 13	1066.03693	R(113-,24)co	(113 ⁻ ,25) ^{co}	\rightarrow	[(113 ⁻ ,24) ^{co}]		[11]	[39.7691]
				\rightarrow	(122 ⁻ ,24) ^{co}	105.40181	ı	105.4020
				\rightarrow	(022+,24)co	285.24496	ı	285.2447
10R(32) - 48	983.25065	P(132,27)co	(132,26)co	\rightarrow	[(023,27)co]		[11]	[164.1724]
				\rightarrow	(023,26)co	207.11286	Τ	207.1128
10R(46) + 48	990.62123	P(123,24)co	(123,23)co	\rightarrow	[(032,23)co]		[1]	[203.3523]
	, ,			\rightarrow	(032,22)co	240.21218	1	240.2123
				\rightarrow	(014,24)co	123.84435	-	123.8450
				\rightarrow	(014,23)co	161.76189	1	161.7618
				\rightarrow	(014,22)co	198.45711	[[]]c	198.4575

- ^a Pump assignments reported in Ref. [2]; the A[±] component was incorrectly identified for 9R(02) 13 MHz.
- b Values in brackets are predicted.
- $^{\rm c}$ The observed \perp polarization given in Table III is inconsistent with the present assignment.

IR pump, although one point needing to be rechecked is that our assignment for the 198.457 11 cm $^{-1}$ laser line requires \parallel polarization rather than the observed \perp polarization given in Table III.

The most important of the new FIR laser lines, spectroscopically speaking, may turn out to be those pumped by the 10R(46) + 15 MHz and the 10R(52) CO₂ lines. Although they are not yet assigned, we can make some pertinent comments about their origins and significance. The two FIR laser lines observed for each system must be b-type members of the triads of Fig. 3; hence, their difference represents the a-type line L'_a in the excited state according to Fig. 3 and (4). The experimental differences of 36.42370 and 36.42201 cm⁻¹ obtained from Table III for the 10R(46) and 10R(52) pumps are very similar, implying that the J values in the two systems must be identical despite the significant difference in pump wavenumber. The upper-level J value and effective B-value for the L'_a transitions are calculated from (5) to be $J_{upper} =$ 23 and B = 0.7918 cm⁻¹, assuming the FIR laser lines correspond to the L_c and L_b transitions of Fig. 3 for each system. This B-value would be expected for the $v_t = 3$ torsional state of the excited CO-stretching mode. Effective B-values decrease with v_t and are equal to 0.7935 cm⁻¹ for the $v_t = 2$ CO-stretch levels [23]. The further reduction of 0.0017 cm⁻¹ observed here is consistent with the calculated decrease from about 0.804 to 0.802 cm⁻¹ in the ground state B-values in going from $v_t = 2$ to $v_t = 3$; hence, it is likely that the lower levels of the FIR laser transitions for the two systems are $v_t = 3$ CO-stretching levels. The identity of the upper level of the lasing and IR pump transition is not clear, and there is a strong possibility of interaction and perturbation given the greater density of states at this level of excitation. However, the tiny absorption peak observed in our IR spectrum at the position of the 10R(52) CO₂ transition suggests pumping from a high torsional level of the ground state with a small Boltzmann factor. It would be rewarding and of great spectroscopic interest to identify such highly excited levels from the FIR laser emission because direct detection and analysis of such weak pump absorptions in our spectrum will certainly be extremely difficult.

The spectroscopic challenge is even greater for the 10R(46) + 15 MHz pump identification because it is not clear which transitions the FIR laser lines correspond to in Fig. 3. The IR pump wavenumber suggests that the pump absorption must be a P-branch transition, in which case the polarization rules of (3) would require \parallel polarization for lines L_b and L_d and \perp polarization for line L_c . The \perp polarization reported for the 192.293 92 cm⁻¹ line in Table III would then make it the Qbranch transition L_c so that the 155.870 22 cm⁻¹ line would be \mathcal{L}_d and \mathcal{J}_{upper} would equal 22 instead of 23. However, the polarization measurements are incomplete for the 10R(46) + 15 MHz system, so the possibility that the FIR laser lines are the L_b and L_c pair cannot be ruled out, and further complication is added to the search for the assignments. It would thus be useful to return to the 10R(46) system in the future to resolve this uncertainty and to reinvestigate the inconsistency between the observed \perp polarization and the polarization required by the assignment for the 198.457 11 cm^{-1} line pumped by 10R(46) at + 48 MHz offset.

V. CONCLUSION

The present paper, in conjunction with other complementary results in three recent reports [14], [16], [17], serves to

summarize the new FIR laser line observations for CH₃OH and thus to update the previous reviews of FIR laser emission for methanol isotopomers [1]-[4]. In this paper, we have reported six new FIR lines for CH₃OH, and accurate heterodyne frequency measurements for 14 lines, nearly all of high frequency above 100 cm⁻¹. Transition assignments are also presented for nine high-frequency lines in four systems and a partial analysis presented for pairs of lines in two other systems suggesting that the FIR laser emission may be associated with highly excited $v_t = 3$ levels in the CO-stretching state.

With the results of the new study, 662 FIR laser lines have now been observed for the parent ¹²CH₃¹⁶OH species so far. To date, we have met the spectroscopic challenge of identifying this extensive emission approximately halfway, with roughly 50% of the known lines having been assigned. In this regard, the new frequency and offset measurements are extremely important in supplying precise benchmark information about transitions in excited vibrational states against which proposed spectroscopic assignments can be rigorously tested and a full and accurate map of the torsion-rotation energy level manifold built up for the excited states.

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